Phenyliodonium Derivatives from 4-Aminocoumarin and their Reactivity

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Dedicated to the memory of Professor Nicholas Alexandrou

The reaction of 4-aminocoumarin with [hydroxy(tosyloxy)iodo]benzene affords 4-amino-3-phenyliodoniocoumarin tosylate which upon basification is converted to the corresponding stable iodine-nitrogen 1.4-dipole. The reactivity of the latter is briefly discussed.

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Aryliodonium phenolates of the general type 1 constitute an interesting class of iodine-oxygen 1,4-dipoles whose chemistry, as well as of other zwitterionic iodonium compounds, has been reviewed [1].

t_{Ar}
$$\tilde{N}H$$

We have recently described the preparation and explored briefly the reactivity of a new class of iodinenitrogen 1,4-dipoles, *i.e.* 3-aryliodonio-1,4-naphthoquinone-2-imides, 2, resulting from the reaction of 2-aminonaphthoquinone with [hydroxy(tosyloxy)iodo]arenes [2].

Extending our studies to other systems containing the same β -keto-enamino functionality, we find that a related heterocyclic substrate, 4-aminocoumarin, 3, forms analogous stable phenyliodonium derivatives. Treatment of 3 with [hydroxy(tosyloxy)iodo]benzene under mild conditions resulted in the formation of its 3-phenyliodonium tosylate, 4, in high yield; upon basification, this was converted to its conjugated acid, *i.e.* the stable 1,4-dipole 5 (Scheme 1).

Scheme 1

The characterization of these new phenyliodonium compounds was based on elemental analysis and spectral data (ir, nmr, ms) which were consistent with the proposed structures. In addition, 5 upon reaction with p-toluenesulfonic acid was converted back to 4. Attempts to explore the reactivity of 5 (photochemical and thermal reactions with alkenes and alkynes) were not successful because isomerization prevailed, with formation of 3-iodo-4-phenylaminocoumarin, 6. Indeed, this rearrangement which has been observed in several analogous 1,4-iodonium zwitterions derived from β -diketones [1], phenols [3,4] or enaminones [2], proceeded quantitatively in refluxing acetonitrile. Most likely, it constitutes another example of a Smiles-type rearrangement.

Deiodination of 6 was effected quantitatively upon treatment with PdCl₂(PPh₃)₂ in triethylamine and tetrahydrofuran. Unexpectedly, the resulting 4-phenylaminocoumarin, 7, did not furnish a stable iodonium derivative similar to 4, when it was treated with [hydroxy(tosyloxy)iodo]benzene, but it was converted directly to 3-tosyloxy-4-phenylaminocoumarin, 9, through the intermediacy of the tosylate 8 (Scheme 2). This reaction is a further demonstration of the superleaving group ability of phenyliodonio group [1] which enables this substitution with such a weak nucleophile under so mild conditions.

Scheme 2

Interestingly enough, when 4-phenylaminocoumarin, 7, was treated with (diacetoxyiodo)benzene, 4-diphenylamino-3-iodocoumarin, 12, was the only isolable product (Scheme 3). The reaction, no doubt, proceeds through the phenyliodonium acetate, 10, which is converted *in situ* to

Scheme 3

the the corresponding dipole, 11; this upon phenyl migration affords compound 12.

In conclusion, we have shown that stable iodine-nitrogen 1,4-dipoles can be obtained from heterocyclic compounds bearing the β -enaminoketone functionality. Our future research will be extended to other similar heterocyclic systems.

EXPERIMENTAL

Melting points were determined on a Kofler hot stage apparatus and are uncorrected. The ir spectra were determined on a Perkin-Elmer 1370 spectrometer. The ¹H nmr spectra were recorded on Bruker AW 80 MHz and AM 300 MHz spectrometers using tetramethylsilane as an internal standard. The ms spectra were recorded on a VG TS-250 instrument.

Compound 3, 4-aminocoumarin, was prepared from the reaction of 4-hydroxycoumarin with ammonium acetate [5].

Preparation of 4-Amino-3-phenyliodoniocournarin Tosylate (4).

[Hydroxy(tosyloxy)iodo]benzene (1.96 g, 5 mmoles) was added to a stirring suspension of 4-aminocoumarin 3 (0.805 g, 5 mmoles) in dichloromethane (30 ml). After one hour at room temperature the resulting white solid was filtered, washed repeatedly with dichloromethane and dried *in vacuo* to yield 2.542 g (95%) of the tosylate 4, mp 194-196°; ir (nujol): v_{max} 3340, 3210, 1680, 1645, 1180, 1160 cm⁻¹; ¹H nmr (80 MHz) (deuteriochloroform/trifluoroacetic acid): δ 8.36-7.13 (m, 13H), 2.39 (s, 3H); ms: m/z 363 (M⁺-TsOH, 60), 331 (M⁺-PhI, 58), 204 (92), 176 (100), 155 (61).

Anal. Calcd. for $C_{22}H_{18}INO_5S$: C, 49.36; H, 3.39; N, 2.62. Found: C, 49.39; H, 3.69; N, 2.48.

Preparation of 3-Phenyliodoniocoumarin-4-imide (5).

A solution of aqueous sodium hydroxide (8 ml, 2%, 4 mmoles) was added to the tosylate 4 (1.07 g, 2 mmoles). After half an hour stirring at room temperature the resulting yellow solid was filtered, washed successively with water and diethyl ether and dried in vacuo to yield 0.668 g (92%) of the dipole 5, mp 93-95°; ir (nujol): v_{max} 3450, 1620, 1595 cm⁻¹; ¹H nmr (80 MHz) (deuteriochloroform/trifluoroacetic acid): δ 8.32-7.31 (m, aromatic H); ms: m/z 363 (M⁺, 44), 236 (57), 170 (100), 77 (75).

Anal. Calcd. for $C_{15}H_{10}INO_2$: C, 49.61; H, 2.78; N, 3.86. Found: C, 49.37; H, 2.61; N, 3.68.

Preparation of 3-Iodo-4-phenylaminocoumarin (6).

A suspension of the dipole 5 (0.363 g, 1 mmole) in acetonitrile (15 ml) was refluxed for three hours. The resulting solution was evaporated to dryness and the remaining crude solid was recrystallized from ethanol to afford 0.29 g (80%) of coumarin 6, mp 181-182°; ir (nujol): v_{max} 3300, 1675, 1585 cm⁻¹; ¹H nmr (80 MHz) (deuteriochloroform): δ 7.69-6.73 (m, aromatic H); ms: m/z 363 (M⁺, 92), 236 (56), 77 (100).

Anal. Calcd. for $C_{15}H_{10}INO_2$: C, 49.61; H, 2.78; N, 3.86. Found: C, 49.78; H, 2.90; N, 3.85.

Deiodination of 6 to 4-Phenylaminocoumarin (7).

A catalytic amount of $PdCl_2(PPh_3)_2$ (0.035 g, 0.05 mmole) was added to a solution of iodocoumarin 6 (0.363 g, 1 mmole) in tetrahydrofuran (10 ml) and triethylamine (5 ml) and the mixture was stirred at room temperature until the disappearance of 6 (3 hours, monitored by tlc). After removal of the solvent, the residue was chromatographed on column (silica gel, hexaneethyl acetate 2:1) to yield 4-phenylaminocoumarin 7 (0.178 g, 75%), mp 266-267° (ethanol) (lit [6] mp 267-268°); ir (nujol): v_{max} 3280, 1655, 1610, 1585 cm⁻¹; ¹H nmr (80 MHz) (deuteriochloroform/deuterated dimethyl sulfoxide): δ 8.76 (s br, 1H, NH), 8.20-7.20 (m, 9H), 5.54 (s, 1H, H-3).

Compound 6 was in all respects identical to 4-phenylaminocoumarin prepared from 4-hydroxycoumarin and aniline, according to the literature method [6].

Preparation of 3-Tosyloxy-4-anilinocoumarin (9).

[Hydroxy(tosyloxy)iodo]benzene (0.784 g, 2 mmoles) was added to a stirring suspension of 4-phenylaminocoumarin 6 (0.474 g, 2 mmoles) in chloroform (15 ml). After 24 hours at room temperature the solvent was removed and the residue was chromatographed on column (silica gel, hexane-ethyl acetate from 10:1 to 2:1). Iodobenzene was eluted first. The second fraction was recrystallized from ethanol to afford 0.245 g of coumarin 9 (67% yield, based on reacted 6), mp 196-198°; ir (nujol): v_{max} 3300, 1670, 1170 cm⁻¹; ¹H nmr (300 MHz) (deuteriochloroform): δ 7.97 (d, J = 8.4 Hz, 2H), 7.46 (m, 1H), 7.24-7.39 (m, 7H), 7.20 (m, 1H), 7.07 (d, J = 8.4 Hz, 2H), 7.01 (m, 1H), 2.44 (s, 3H); ¹³C nmr (75 MHz) (deuteriochloroform): δ 21.8, 114.6, 117.6, 123.0, 123.5, 125.3, 125.9, 126.3, 128.9, 129.4, 129.7, 131.9, 132.3, 140.3, 145.1, 146.0, 153.0, 159.0.

Anal. Calcd. for $C_{22}H_{17}NO_5S$: C, 64.85; H, 4.21; N, 3.44. Found: C, 64.68; H, 4.31; N, 3.58.

Unreacted 4-phenylaminocoumarin (0.261 g) was the third fraction.

Preparation of 4-Diphenylamino-3-iodocoumarin (12).

(Diacetoxy)iodobenzene (0.966 g, 3 mmoles) was added to a stirring suspension of 4-phenylaminocoumarin 7 (0.237 g, 1 mmole) in dichloromethane (15 ml). After one day at room temperature the resulting solution was evaporated and the residue was chromatographed on column (silica gel, hexane-ethyl acetate 10:1). After a small amount of iodobenzene, 4-diphenylamino-3-iodocoumarin 12 (0.307 g, 70%) was eluted and recrystallized from ethanol, yellow crystals, mp 247-248°; ir (nujol): v_{max} 1695, 1580 cm⁻¹; ¹H nmr (80 MHz) (deuteriochloroform): δ 7.54-7.00 (m, aromatic H); ms: m/z 439 (M+ 28), 312 (M+-I, 81), 77 (100).

Anal. Calcd. for $C_{21}H_{14}INO_2$: C, 57.42; H, 3.21; N, 3.19. Found: C, 57.41; H, 3.11; N, 3.20.

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